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Quiz & Worksheet - Covalent Chemical Bonds

1. If a silicon atom has four valence electrons, what is the maximum number of bonds it can form?

- a. 2
- b. 4
- c. 1
- d. 8

2. Which of the following statements are true?

- I. Noble gases complete their valence level by sharing electrons.
- II. Covalent bonds are weak and easily broken.
- III. Covalent bonds allow us to make predictions on how the molecule will look.

- a. I
- b. II
- c. III
- d. I and III

3. If a carbon atom has four valence electrons, which of the following statements are true?

- I. Carbon can form a triple covalent bond with another carbon atom.
- II. A carbon atom cannot form two double bonds simultaneously.
- III. A carbon atom can form four single covalent bonds simultaneously.

- a. I
- b. II
- c. III
- d. I and III

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