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[Section 72 Types Reactions Answers](#)

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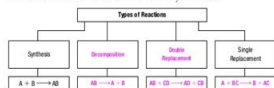
Chapter 7 Chemical Reactions

**Section 7.2 Types of Reactions**  
(pages 199-205)

This section discusses how chemical reactions are classified into different types.

**Reading Strategy** (page 199)

**Previewing** Skim the section and begin a concept map like the one below that identifies types of reactions with a general form. As you read, add the general form of each type of reaction. For more information on this Reading Strategy see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.



**Classifying Reactions** (pages 199-204)

- Name five general types of chemical reactions. *Synthesis, decomposition, single replacement, double replacement, and combustion.*
- Circle the letter of each equation that represents a synthesis reaction.
  - $2Na + Cl_2 \rightarrow 2NaCl$
  - $2NaCl \rightarrow 2Na + Cl_2$
  - $2H_2O \rightarrow 2H_2 + O_2$
  - $2H_2 + O_2 \rightarrow 2H_2O$
- Is the following sentence true or false? A decomposition reaction is the opposite of a synthesis reaction. *True*
- Write the equation for the decomposition of calcium carbonate into calcium oxide and carbon dioxide.  *$CaCO_3 \rightarrow CaO + CO_2$*
- Circle the letter of the correct answer. Copper reacts with silver nitrate in a single-replacement reaction. What are the products of this reaction?
  - copper(II) nitrate and silver oxide
  - copper(II) nitrate and silver
  - copper(II) oxide and silver nitrate
  - copper, nitrogen, and silver oxide